

Chemistry 1 Final Exam Study Guide

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A calorie is the amount of energy it takes to heat 1 g of water by 1 degree celcius. A caloire is equal to 1,000 calories

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CH₃COOH reaction any chemical change in which one species is

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oxidized (loses electrons) and another species is reduced (gains electrons) also called redox reaction precipitation reaction A reaction in which two solutions react to form an insoluble solid, which separates from the solution.

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Chemistry I Final Exam Study Guide. Terms in this set (32) the symbols for the elements combined in a compound with subscripts showing the smallest whole-number mole ratio of the different atoms in the compound. a formula the types and numbers of atoms combined in a single molecule of a molecular compound.

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Chemistry 11 Final Exam Study Guide Chemistry 11 - Final Exam Study Guide Page 15 When Electronegativities of bonding atoms

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are the same (as they are in diatomic molecules) or close to the same, they SHARE electrons. Bonds formed when atoms share electrons are called Covalent Bonds .

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The final exam will contain questions VERY similar to the ones that you saw on your Unit Tests. If you still have your Unit Tests, it is highly recommended that you use those to review for the final exam in addition to these flashcards.

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Instructions. Choose your answer to the question and click 'Continue' to see how you did. Then click 'Next Question' to answer the next question. When you have completed the free practice test, click 'View Results' to see your results.

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The American Chemical Society sells study guides including the General Chemistry Study Guide (ISBN: 0-9708042-0-2). The first thing you need to do is purchase the ACS official study guide. The book is just over 100 pages long and provides example questions along with explanations for the correct answer.

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given a paper reference table but the exam is on the computer! You must print the "Fall 2018 Final Exam Review" on your own!

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CHEM 443 (Physical Chemistry I) Application of the principles and methods of physics and mathematics to the quantitative study of chemistry, focusing on chemical thermodynamics, electrochemistry, and elementary concepts in chemical transport phenomena.

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