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Gases with high pressure exert more force than gas with low pressure. The SI unit of pressure is the pascal (Symbol Pa). The pascal is equal to the force of 1 newton per square meter.

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Page ID 54606; No headers. The study of gases allows us to understand the behavior of matter at its simplest: individual particles, acting independently, almost completely uncomplicated by interactions and interferences between each other. This knowledge of gases will serve as the pathway to our understanding of the far

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more complicated condensed phases ...

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drugs and excipients, frequently exhibit physicochemical properties that. Using balloons to illustrate the process, Eureka! shows how, when matter gets hot, it's molecules go ...

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Characteristics of Gases •All substances

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have three phases: solid, liquid and gas.

- Substances that are liquids or solids under ordinary conditions may also exist as gases.
- These are often referred to as vapors.
- Many of the properties of gases differ from those of solids and liquids:
- Gases are highly compressible and ...

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aumackenzie0817 . Terms in this set

(32) What are some properties of gases?

They have mass, low densities, they are easily compressed, completely fill their container, can move through each other

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rapidly, and exert pressure. What is diffusion? The movement of one substance through ...

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Gases will expand to the greatest extent possible, resulting in low mass per unit volume ratios. Compressibility is also linked to expansion and the indefinite shape of the gas, essentially suggesting

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that the distance between particles can be reduced if pressure is increased. Diffusion and effusion are both linked to the movement of gases. Diffusion means that gases can spread out and mix ...

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state of matter is the simplest and best-understood state of matter. You inhale approximately 8,500 L of air each day.

This amounts to about 25 lbs of air.

Breathing is a three-step process: inhaling, gas exchange with the circulatory system, and exhaling.

Approximately 80% pressure differences created by your body allow you to

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breathe. Nearly the Earth's ...

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dipole-dipole forces which are stronger than dispersion forces Apply the relationship between bond energy and length of single, double, and triple bonds (conceptual, no ...

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